

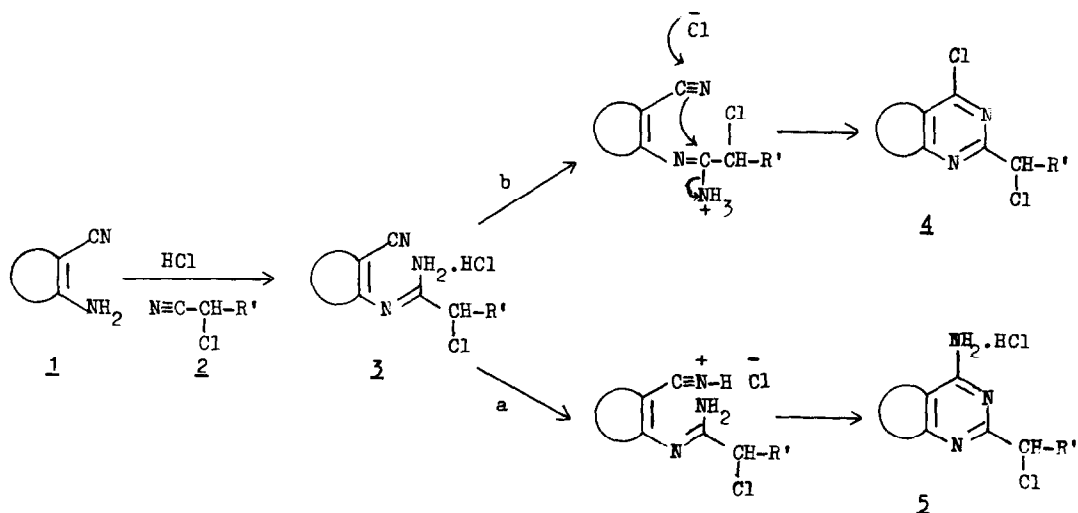
REACTION OF NITRILES UNDER ACIDIC CONDITIONS:
 A NOVEL, DIRECT FORMATION OF CONDENSED 4-CHLOROPYRIMIDINES

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Summary: An unusual formation of condensed 4-chloropyrimidines in the reaction of o-aminonitriles with halogenoacetonitriles in the presence of dry hydrogen chloride gas is described.

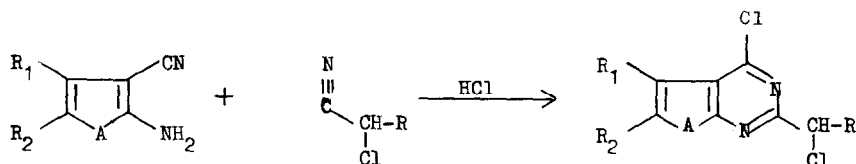
We have earlier described^{1,2} a facile one-pot synthesis of fused pyrimidines by the reaction of nitriles with o-aminocarbonyl compounds in the presence of dry hydrogen chloride gas. We have now observed the unusual direct formation of condensed 4-chloropyrimidines **4** instead of the normally expected 4-aminopyrimidines **5** in the condensation of o-aminonitriles **1** with certain halogenoacetonitriles **2**. Thus, passing a stream of dry hydrogen chloride gas through a mixture of the reactants in dioxane at 5-10°C for 8 hours and the usual work-up of the reaction mixture yielded the condensed 4-chloropyrimidines presented in Table I.

The exclusive formation of 4-chloropyrimidines **4** over that of 4-aminopyrimidines **5** can be rationalized as occurring through the preferential cyclization of the initially formed amidine intermediate **3** by the pathway 'b' which may be attributed to the increased electrophilicity of the amidine carbon and the decreased nucleophilicity of the amidine nitrogen in the intermediate **3**, as a consequence of the electron withdrawing effect of the chlorine atom (Scheme I).



Scheme I

Table I



R ₁	R ₂	A	R	Mol. Formula ^a	Mass ^b M ⁺	M.P. °C	Yield %
H	H	CH=CH	H	C ₉ H ₆ N ₂ Cl ₂	212	100-102 ^c	85
H	H	CH=CH	Cl	C ₉ H ₅ N ₂ Cl ₃	246	135-137	87
	-(CH ₂) ₄ -	S	H	C ₁₁ H ₁₀ N ₂ SCl ₂	272	98-100	69
	-(CH ₂) ₄ -	S	Cl	C ₁₁ H ₉ N ₂ SCl ₃	306	118-120	78
CH ₃	CH ₃	S	H	C ₉ H ₈ N ₂ SCl ₂	246	141-143	73
C ₆ H ₅	C ₆ H ₅	O	H	C ₁₉ H ₁₂ N ₂ OCl ₂	354	120-122	58

a) All the compounds were recrystallized from hexane. All the compounds gave correct analysis. b) Mass spectra were recorded on a Varian-Atlas CH-7 mass spectrometer at 70 eV ionizing beam using direct insertion probe. c) Ref. No. 3; Reported M.P. 95-96°C.

This novel cyclization bears a formal resemblance to the cyclization of α,ω -dinitriles and analogues to halogenoazines⁴ and of *o*-cyanophenyl isocyanates to 4-halogenoquinazolin-2-ones⁵, catalyzed by halogen acids. A detailed paper on the mechanism and the scope of this reaction, as well as, the factors influencing the formation of 4-chloro- and 4-amino- condensed pyrimidines in the condensation of electronegatively substituted nitrile derivatives with *o*-aminonitriles will be published shortly.

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REFERENCES

1. K.G. Dave, C.J. Shishoo, M.B. Devani, R. Kalyanaraman, S. Ananthan, G.V. Ullas, and V.S. Bhaddi, *J. Heterocycl. Chem.*, **17**, 1497(1980).
2. C.J. Shishoo, M.B. Devani, M.D. Karvekar, G.V. Ullas, S. Ananthan, V.S. Bhaddi, R.B. Patel, and T.P. Gandhi, *Indian J. Chem.*, **21B**, 666(1982).
3. H. Breuer, *Tetrahedron Letters*, 1935(1976).
4. F. Johnson and R. Madronero, *Adv. Heterocycl. Chem.*, **6**, 95(1966).
5. G. Simchen, G. Entenmann, and R. Zowdler, *Angew. Chem. Int. Ed.*, **9**, 523(1970).

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